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2010 J. Phys.: Conf. Ser. 207 012019

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doi:10.1088/1742-6596/207/1/012019

# AC Plasma Induced Modifications in Sb<sub>2</sub>S<sub>3</sub> Thin Films

M. Calixto-Rodriguez<sup>1</sup>, F. Castillo<sup>2,\*</sup>, H. Martínez<sup>1</sup>, Y. Peña<sup>3</sup>, and A. Sanchez-Juarez<sup>4</sup>

**Abstract**.  $Sb_2S_3$  thin films, deposited by the chemical bath deposition method, were treated with  $N_2$  plasma at 3.0 Torr during several minutes. The as-prepared  $Sb_2S_3$  thin films and films treated with  $N_2$  plasma have been characterized using several techniques. X-ray diffraction studies have shown that plasma treatment induced recrystallization on the as-prepared  $Sb_2S_3$  thin films. The band gap values decreased from 2.37 to 1.82 eV after plasma treatment, and the electrical conductivity increased from  $10^{-9}$  to  $10^{-7}$  ( $\Omega$  cm)<sup>-1</sup> due to the annealing effect.

#### 1. Introduction

Antimony sulfide  $(Sb_2S_3)$  has technological applications as target material of television cameras [1], microwave [2], switching [3], and optoelectronic devices [4]. Recently, considerable attention has been given to the preparation and characterization of  $Sb_2S_3$  thin films for use in photovoltaic devices [5-8]. Chemical bath deposition is a very important method used to grow metal chalcogenide materials in thin film form for solar cell applications, due to the simplicity and non-expensive of the technique.

Plasma treatment is an interesting technological method to modify the structural [9] and surface [10] properties of the materials. This paper deals with  $N_2$  plasma treatment used to modify the structural, optical, and electrical properties of the as-prepared samples of  $Sb_2S_3$  thin films obtained by the chemical bath deposition method.

## 2. Experimental details

Antimony sulfide thin films were prepared using the chemical bath deposition technique, similarly to that already reported in [11]. The bath was prepared using antimony trichloride, SbCl<sub>3</sub> and sodium thiosulfate, Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub>, and deionized cold (~10°C) water. Microscope glass slides were used as substrates. The substrates were placed vertically in the solution. The deposition was made at ~10 °C for 4 hours without stirring. Thickness of the films was measured with an Alpha Step model 100

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<sup>&</sup>lt;sup>1</sup> Instituto de Ciencias Físicas, Universidad Nacional Autónoma de México, Apartado Postal 48-3, 62210, Cuernavaca, Morelos, México.

<sup>&</sup>lt;sup>2</sup> Instituto de Ciencias Nucleares, Universidad Nacional Autónoma de México, Apartado Postal 70-543, 04510, México D. F., México.

<sup>&</sup>lt;sup>3</sup>Universidad Autónoma de Nuevo León, Facultad de Ciencias Químicas, Pedro de Alba s/n, Cd. Universitaria, San Nicolás de los Garza, N.L.

<sup>&</sup>lt;sup>4</sup>Centro de Investigación en Energía, Universidad Nacional Autónoma de México, Privada Xochicalco s/n Col. Centro, Temixco, Morelos, México; C.P. 62580.

<sup>\*</sup>E-mail: ciro@nucleares.unam.mx

<sup>\*</sup> To whom any correspondence should be addressed.

doi:10.1088/1742-6596/207/1/012019

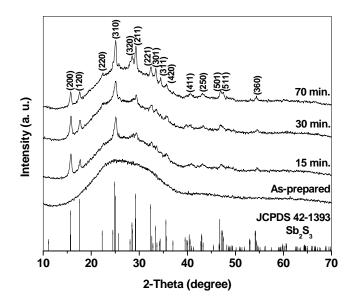
profilometer from Tencor Instruments, the mean value of the film thickness was 240 nm. The asprepared samples of Sb<sub>2</sub>S<sub>3</sub> were treated with N<sub>2</sub> plasma at 3.0 Torr during 15, 30, and 70 minutes.

The experimental apparatus and technique to generate the pulsed plasma was recently reported [12]. The discharge power supply was maintained at an output of 300 Volts and a current of 0.36 A. The glow discharges were monitored by plasma emission spectroscopy using an Ocean Optics Inc. Spectrometer Model HR2000CG-UV-NIR. X-ray diffraction (XRD) patterns were recorded on a Rigaku D-Max X-ray diffractometer using Cu-K $_{\alpha}$  radiation ( $\lambda$  = 1.5406 Å). The optical transmittance and specular reflectance spectra of the samples were measured with a spectrophotometer Shimadzu model UV-1601PC. For the electrical measurements, current vs. time data were recorded on an automated system using a Keithley 619 electrometer and a Keithley 230 programmable voltage source.

## 3. Results and discussion

## 3.1 Structural analysis

The XRD patterns for as-prepared  $Sb_2S_3$  thin films as well as films treated with  $N_2$  plasma are shown in figure 1. It can be seen that no diffraction peaks can be identified on the pattern of the as-prepared film. It is possible that the material is growing with nanometric grains. The diffraction patterns corresponding to films treated with  $N_2$  plasmas during 15, 30, and 70 minutes match well the standard for  $Sb_2S_3$  (JCPDS 42-1393) which has an orthorhombic structure. Films treated with  $N_2$  plasma showed a preferential orientation along the (310) direction. The mean value of the crystallite size for  $Sb_2S_3$  films treated with  $N_2$  plasma was calculated for the (310) diffraction peak, using the Scherrer formula  $D = (0.9\lambda)/(\beta\cos\theta)$ , where D is the diameter of crystallites,  $\lambda$  is the wavelength of Cu- $K_\alpha$  line,  $\beta$  is full width at half maximum (FWHM) in radians and  $\theta$  is Bragg's angle. It was found that the value increased from 15.3 nm to 17.8 nm with the increasing in time to the  $N_2$  plasma treatment. This may be due to the recrystallization of the films during the plasma process.



**Figure 1.** XRD patterns for as-prepared Sb<sub>2</sub>S<sub>3</sub> thin films as well as films treated with N<sub>2</sub> plasma.

## 3.2 Optical emission spectroscopy analysis

Optical emission spectroscopy (OES) measurements were obtained for N<sub>2</sub> glow discharge plasma (figure 2). This fact allowed the analysis of the most luminous area that corresponds to the negative

Journal of Physics: Conference Series 207 (2010) 012019

doi:10.1088/1742-6596/207/1/012019

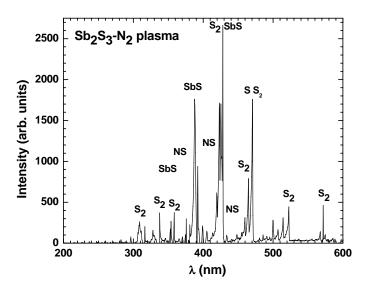
glow near the cathode dark space (it has been subtracted the intensities of the bands coming from the  $N_2$  plasma). The identified species removed from the thin films into the plasma were  $S_2$ , NS,  $S_2$  and  $S_2$  by  $S_3$ . The majority of the species are vibrationally excited at a fundamental level. There is a flux of material into the plasma that can be deposited somewhere in the chamber and some back on the surface from which it came (in a modified form). By using OES and observing both  $S_2$ ,  $S_3$  by  $S_3$  and  $S_3$  be neutrals, it is possible to assume that there is a possible mechanism by which  $S_3$  thin film ionization occurs in plasma treatment: electron impact ionization.

$$Sb_2S_3 + e \Rightarrow S_2 + Sb_2S + e \tag{1}$$

$$Sb_2S_3 + N \implies NS + Sb_2S_2 + e \tag{2}$$

$$Sb_2S_3 + e \Rightarrow Sb + SbS_3 + e$$
 (3)

$$Sb_2S_3 + e \Rightarrow SbS + SbS_2 + e$$
 (4)



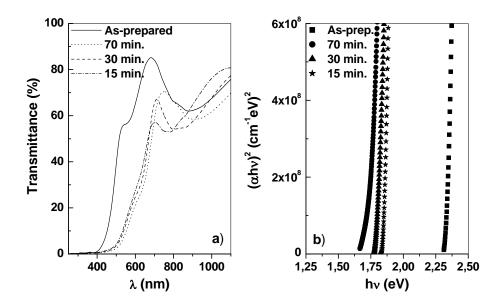
**Figure 2.** Typical OES spectrum for Sb<sub>2</sub>S<sub>3</sub> thin film-N<sub>2</sub> plasma interaction.

## 3.3 Optical and electrical measurements

The behaviour of the transmittance,  $T(\lambda)$ , for as-prepared  $Sb_2S_3$  thin films as well as films treated with  $N_2$  plasma at 3.0 Torr during 15, 30, and 70 minutes are presented in figure 3a. Samples after plasma treatments show a decrease in the optical transmission due to the structural changes. Band gap,  $E_g$ , of the samples was determined using the  $(\alpha h v)^2$  versus h v graph by extrapolating the linear portion of the graph to the h v axis (figure 3b). The as-prepared  $Sb_2S_3$  thin film had a band gap of 2.33 eV. As the exposition time to the  $N_2$  plasma increased, the band gap decreased. The  $E_g$  values for films treated with  $N_2$  plasma during 15, 30, and 70 minutes were: 1.85 eV, 1.80 eV, and 1.73 eV, respectively. This decrease might be due to the increasing in the particle size as observed in the XRD studies.

The electrical conductivity value for the as-prepared thin films was  $1.0 \times 10^{-9}$  ( $\Omega$  cm)<sup>-1</sup>. The values were  $\sim 2 \times 10^{-9}$ ,  $3 \times 10^{-8}$ , and  $1 \times 10^{-7}$  ( $\Omega$  cm)<sup>-1</sup> for samples treated with N<sub>2</sub> plasma during 15, 30, and 70 minutes, respectively. These results are in agreement with those reported in references [11,13] for Sb<sub>2</sub>S<sub>3</sub> thin films deposited by chemical bath deposition and thermal annealed in N<sub>2</sub>.

doi:10.1088/1742-6596/207/1/012019



**Figure 3.**  $T(\lambda)$  and  $(\alpha h v)^2$  vs h v plots for as-prepared Sb<sub>2</sub>S<sub>3</sub> film and films treated with N<sub>2</sub> plasma.

## 4. Conclusions

 $N_2$  Plasma treatments induced recrystallization was observed in  $Sb_2S_3$  thin films due to annealing effect of plasma process. Films treated in  $N_2$  plasma showed optical and electrical properties similar to that obtained through a conventional thermal treatment.

#### Acknowledgments

The authors are thankful to Maria Luisa Ramón for the XRD measurements, José Ortega (CIE-UNAM), A. Bustos, A. González, R. Bustos (ICF-UNAM), and José Rangel (ICN-UNAM) for technical assistance. One of the authors (MCR) acknowledges the financial support received from DGAPA-UNAM. This research was partially sponsored by DGAPA IN-105707-3, and CONACyT 41072-F.

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